

CHEMICAL TRANSGLYCOSYLATION OF OCTOSYL ACID

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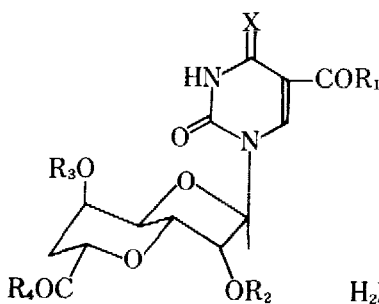
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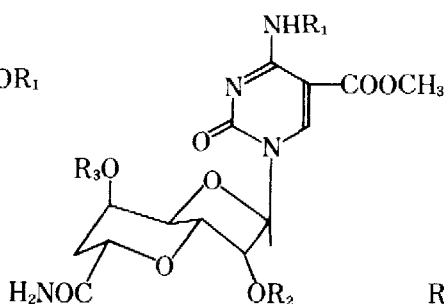
In a previous communication,¹ we have reported the isolation and the structure of anhydrooctose uronic acid nucleosides, octosyl acids, which are metabolites of *Streptomyces cacaoi* var *asoensis*. Since these nucleosides were regarded as carboanalogs of 3',5'-cyclic nucleotides, it was especially interesting to convert the 5-substituted uracil base of octosyl acids into adenine to prepare an analog of biologically important cyclic AMP. Transglycosylation of pyrimidine nucleosides to purine nucleosides was described before by Miyaki *et al.*² However, since their procedure was not mild enough to apply to the labile sugar skeleton of octosyl acids,¹ we studied a more efficient method, utilizing N⁶-benzoyl,9-bis(trimethylsilyl)adenine and trimethylsilyl perchlorate as a catalyst in



1 $R_1=R_4=OH$, $R_2=R_3=H$, $X=O$

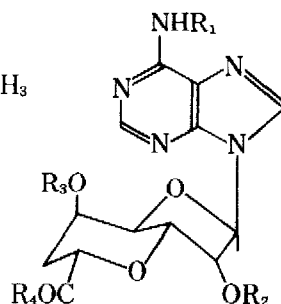
2 $R_1=R_4=OMe$, $R_2=R_3=Ac$, $X=O$

3 $R_1=R_4=OMe$, $R_2=R_3=Ac$, $X=S$



4 $R_1=R_2=R_3=H$

5 $R_1=R_2=R_3=Ac$



6 $R_1=Bz$, $R_2=R_3=Ac$, $R_4=NH_2$

7 $R_1=R_2=R_3=H$, $R_4=OH$

dichloroethane—acetonitrile. This modification is based on a recent report by Vorbruggen and Krolkiewicz³ concerning new catalysts for nucleoside synthesis.

Octosyl acid A (1) was esterified followed by acetylation to give the dimethylester diacetate (2), mp 229–231°. Thiation of 2 with phosphorous pentasulfide in dioxane⁴ afforded the 4-thiouracil derivative (3), mp 206–208°, which on treatment with 20% ammoniacal methanol at 100° yielded a cytosine derivative (4), mp 290–293° (dec), $C_{14}H_{18}N_4O_8 \cdot H_2O^5$ in good yield. Acetylation of 4 yielded triacetyluronamide (5), mp 164–171°. Mass spectrum (70 eV, LKB 9000S instrument): M, m/e 496 (0.5% relative intensity); sugar, 286 (32); base + CH_2O ,^{6,7} 240 (14); base + 2H, 212 (94), sugar - HOAc - Ac, 183 (59); base + 2H - CH_2CO , 170 (100), sugar - 2HOAc, 166 (94).

Transglycosylation to form an adenine nucleoside was performed as follows. To an acetonitrile solution (10 ml) of 5 (1.0 mmole) was added a solution of *N*⁶-benzoyl,9-bis(trimethylsilyl)adenine⁸ (2.0 mmoles) in 4.5 ml of dichloroethane, followed by addition of 1.2 mmoles of trimethylsilyl perchlorate⁹ in 2 ml of dichloroethane. The solution was refluxed for 20 hr. After work-up of the reaction mixture with a cold dichloroethane—aqueous sodium bicarbonate, followed by silica gel chromatography in the solvent, chloroform—methanol (50/1), an adenine derivative, 9- β -(3,7-anhydro-6-deoxy-2,5-*O*-diacetyl-D-glycero-D-*allo*-octofuranosyluronic acid)-*N*⁶-benzoyladenine (6) was obtained in 60% yield as a crystalline powder. Uv max (EtOH): 231, 279 nm (ϵ 13,400, 19,200), $[\alpha]_D^{20} = +10.0^\circ$ (c 1.24, $CHCl_3$). Mass spectrum (70 eV) M, m/e 524 (2.7% relative intensity), M - H - CO, 495 (16); sugar, 286 (18); base + CH_2O , 268 (9); base + 2H, 240 (45), 210 (28); sugar - HOAc - Ac, 183 (6.2), sugar - 2HOAc, 166 (32); C_6H_5CO , m/e 105 (100). Pmr ($CDCl_3$) δ 1.86 (m, 1, H-6'a), 2.56 (m, 1, H-6'e), 1.99, 2.21 (s, 3H each, CH_3COO), 4.01 (q, 1, H-4'), 4.53 (q, 1, H-7'), 4.97 (q, 1, H-3'), 5.68 (broad, 1, H-5'), 5.71 (d, 1, H-2'), 6.03 (s, 1, H-1'), 6.12, 6.43 (broad, 1H each, $CONH_2$), 7.4–7.7, 7.9–8.1 (m, 5, benzoyl H), 8.07, 8.78 (s, 1H each, H-2 and H-8), 9.31 (broad s, 1, 6-NHBz); $J_{1',2'} = 0$, $J_{2',3'} = 6.0$, $J_{3',4'} = 10.0$, $J_{4',5'} = 3.0$, $J_{5',6'e} = J_{5',6'a} = \sim 3$, $J_{6'a,7'} = 12.5$, $J_{6'e,7'} = \sim 3$ Hz.

Treatment of 6 with 0.2 *N* sodium methoxide in methanol, followed by 0.5 *N* sodium hydroxide afforded a crystalline free nucleoside 7 in 80% yield; mp >280°, $C_{13}H_{15}N_5O_5 \cdot H_2O$.⁵ Uv max: H_2O , 260 nm (ϵ 12,700); 0.05 *N* HCl, 257 (12,500); 0.05 *N* NaOH, 260 (13,000) $[\alpha]_D^{20} = -35.0^\circ$ (c 0.3, *N* NH_4OH). CD min (H_2O): $[\theta]_{260} = -8830$. Pmr (1% ND_4OD) 6.30 (s, 1, H-1'), 8.36, 8.42 (s, 1H each, H-2 and H-8).

The mass spectrum of the final product 7 after trimethylsilylation¹⁰ is shown in Figure 1. Assignments of the principal peaks follow those for silylated nucleosides¹¹ and confirm the presence of adenine and octosyl acid sugar moieties. M (tetrasilyl derivative), m/e 625, M - CH_3 , 610, sugar - H, 418; sugar - CO, 390; 418 - TMSOH, 328, base + C_2H_3OTMS , 322; 390 - TMSOH, 300;

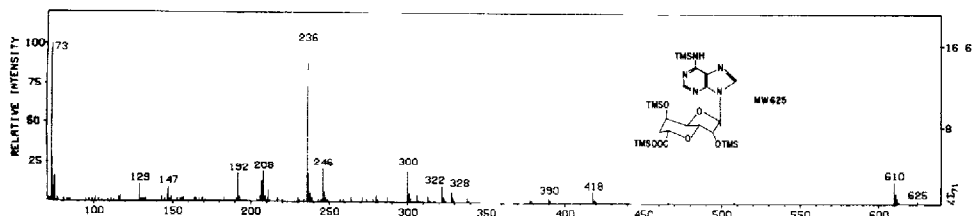


Figure 1 Mass spectrum of the trimethylsilyl derivative of 7

$C_4H_4O_3(TMS)_2$, 246, base + CH_2O , 236, base + 2H, 208; base + H - CH_3 , 192; $SiMe_3$, 73.

A sharp singlet of anomeric protons in pmr of 6 and 7 proves the intact 3,7-anhydrooctofuranose uronic acid skeleton¹ as well as β -orientation of the nucleoside bond.^{1,2} No appreciable amount of α -anomer was observed. The negative Cotton effect observed in 7 indicates the *anti* conformation.

This improved transglycosylation reaction may prove to be a versatile method for the interconversion of purine, pyrimidine, and analogous *N*-nucleosides. This line of study is currently in progress in this laboratory. Biological activity of 7 is also under investigation.

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References and Notes

1. K. Isono, P. F. Crain, and J. A. McCloskey, *J. Am. Chem. Soc.*, **97**, 943 (1975).
2. M. Miyaki, A. Saito, and B. Shimizu, *Chem. Pharm. Bull.*, **18**, 2457 (1970).

3. H. Vorbruggen and K. Krolikiewicz, Angew. Chem., 87, 417 (1975).
4. E. A. Falco, B. A. Otter, and J. J. Fox, J. Org. Chem., 35, 2326 (1970).
5. Satisfactory elemental analysis was obtained as a monohydrate.
6. K. Blemann and J. A. McCloskey, J. Am. Chem. Soc., 84, 2005 (1962).
7. S. J. Shaw, D. M. Desiderio, K. Tsuboyama, and J. A. McCloskey, J. Am. Chem. Soc., 92, 2510 (1970).
8. T. Nishimura and I. Iwai, Chem. Pharm. Bull., 12, 352 (1964).
9. U. Wannagat and W. Liehr, Angew. Chem., 69, 783 (1957), H. C. Marsmann and H. G. Horn, Z. Naturforsch., 27b, 1448 (1973). *Cautron!* Trimethylsilyl perchlorate is very explosive. Use of trimethylsilyl trifluoromethanesulfonate gave almost the same result.
10. The trimethylsilyl (TMS) derivative of 7 was prepared by a method similar to that described in: S. E. Hattox and J. A. McCloskey, Anal. Chem., 46, 1378 (1974)
11. J. A. McCloskey, A. M. Lawson, K. Tsuboyama, P. M. Krueger, and R. N. Stillwell, J. Am. Chem. Soc., 90, 4182 (1968).
12. R. H. Lemieux and J. W. Lown, Can. J. Chem., 41, 889 (1962).